

# Prentice Hall Chemistry Laboratory Manual

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## pH Wikipedia

December 8th, 2018 - In chemistry pH is a logarithmic scale used to specify the acidity or basicity of an aqueous solution It is approximately the negative of the base 10 logarithm of the molar concentration measured in units of moles per liter of hydrogen ions More precisely it is the negative of the base 10 logarithm of the activity of the hydrogen ion

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